

Gas Laws Study Guide

Test Format:

- Multiple choice and problems

Items that will be provided for the test (read important notes at the bottom):

- Gas law equations*
- The value of R
- Pressure conversion factors

* I will provide the equations on the test. You are responsible for knowing which laws the equations describe (ex: I will give you $P_1 \times V_1 = P_2 \times V_2$ and you must know that this is Boyle's Law).

- What is the name of the model we currently use to explain the behavior of gases? **kinetic molecular theory (KMT).**
- Define the kinetic molecular theory (KMT). **Gases are in constant motion with elastic collisions**
- The KMT makes many assumptions about the Size, motion, and energy of gas particles.
- According to the KMT, do gas particles experience attractive or repulsive forces? Explain why or why not. **Neither- they are too far apart**
- Completely describe the motion of gas particles according to the KMT. **Constant, Random, Straight paths, different velocity**
- Define elastic collision. **No kinetic energy is lost- just transferred between particles**
- What determines the kinetic energy of a particle? **Mass and velocity**
- In a sample of a single gas, do all particles have the same kinetic energy? Explain why or why not. **No — they have a different velocity**
- Define temperature. **Measurement of kinetic energy**
- Explain why gases have a low density. **Lots of space between particles (large volume)**
- Why are gases compressible? **Lots of empty space between particles**
- Why do gases expand to fill a container? **No attractive or repulsive forces**
- Why can gas particles flow easily past each other? **No attractive forces**
- Define diffusion. **Movement of one material through another**
- Define effusion. **Gas escapes through tiny openings**
- Describe how diffusion depends on the mass of the particles. **Diffusion is dependent on mass, so less mass more diffusion**
- Use the KMT to explain why a balloon expands as you blow air into it. **Particles are in constant motion and collide with their surroundings expanding the balloon**

Boyle's Law: Answer the following questions about Boyle's Law.

- Describe Boyle's Law in your own words. **As pressure increases, volume decreases**
- What is the formula? **$P_1V_1 = P_2V_2$**
- What are the variables? What are the possible units used in the formula? **Pressure and volume, atm, kPa, mL, L**
- What is constant? **temperature**



- Give an example to help explain the law (you should have some type of visual of your example).
- What would a graph of pressure vs. volume of a gas look like according to Boyle's Law?

Charles' Law: Answer the following questions about Charles' Law.

- Describe Charles' Law in your own words. **Increase temperature, increase volume**
- What is the formula? **=**
- What are the variables? What are the possible units used in the formula? **Temperature and volume (kelvin, mL, L)**
- What is constant? **Pressure**
- What is very important to remember about the temperature variable when using this formula? **It must be in kelvin**
- Give an example to help explain Charles's Law (you should have some type of visual of your example).



- What would a graph of temperature vs. volume of a gas look like according to Charles's Law?

Gay-Lussac's Law: Answer the following questions about Gay-Lussac's Law.

- Describe Gay-Lussac's Law in your own words. **As pressure increases, temperature increases**
- What is the formula? **=**
- What are the variables? What are the possible units used in the formula? **Pressure and temperature (kelvin, atm, kPa, torr)**
- What is constant? **volume**
- What is very important to remember about the temperature variable when using this formula? **It must be in kelvin**
- Give an example to help explain Gay-Lussac's Law (you should have some type of visual of your example).

Study Guide For Content Mastery Gas Laws

M Woodhall



Study Guide For Content Mastery Gas Laws:

Resources in Education ,1992 *Development and Management of Virtual Schools* Catherine Cavanaugh,2004-01-01
Virtual schools are a result of widespread changes in knowledge about learning in available technology and in society Virtual schooling is growing in popularity and will continue to attract students because of the benefits it offers over traditional schooling Stakeholders in virtual schools need information to guide their decisions For the foreseeable future virtual schools will continue to meet diverse student needs and to evolve in response to further change Development and Management of Virtual Schools Issues and Trends brings together knowledge of virtual schools as a reference for scholars and other groups involved in virtual schools The chapters review best practice from concept and development through implementation and evaluation *Study Guide* George Karleskint,1991 Research in Education ,1974 *Teaching Science Creatively in the Secondary Schools* Nathan Seymour Washton,1967 **Comprehensive Respiratory Therapy Exam Preparation Guide (book)** Craig L. Scanlan,Al Heuer,2013-09-09 The Ultimate Review Guide for the CRT RRT and CSE Exams Continuous Up to date NBRC Examination Guidelines and Correlations on Companion Website Comprehensive Respiratory Therapy Exam Preparation Guide Second Edition is a comprehensive study guide for respiratory therapy students and graduates of accredited respiratory therapy education programs who are seeking to take the Certified Respiratory Therapist CRT or Registered Respiratory Therapist RRT credentialing exams from the National Board for Respiratory Care NBRC Comprehensive Respiratory Therapy Exam Preparation Guide Second Edition is reflective of the current CRT RRT and CSE exam matrix and authored by experts who take the credentialing exam annually so you can be confident that the content and format of this guide is current Important Notice The digital edition of this book is missing some of the images or content found in the physical edition **Soviet Education** ,1965 **Proceedings of the 2025 9th International Seminar on Education, Management and Social Sciences (ISEMSS 2025)** Qing Ding,Yixuan Wang,Jianfei Luo,Hongbo Li,2025-10-12 This is an open access book 2025 9th International Seminar on Education Management and Social Sciences ISEMSS 2025 aims to bring together scholars researchers practitioners and policymakers from across the globe to engage in insightful discussions and collaborations on pressing issues in the realms of education management and social sciences As we navigate an increasingly complex and interconnected world this conference offers a vital platform for sharing innovative ideas and best practices that can drive advancements in these critical fields **School Science and Mathematics** ,1925 Daily Graphic Ransford Tetteh,2014-03-11 **Fire Investigator** International Association of Fire Chiefs,International Association of Arson Investigators,National Fire Protection Association,2011-02-14 At head of title International Association of Fire Chiefs International Association of Arson Investigators National Fire Protection Association *Critical Skills and Procedures in Emergency Medicine, An Issue of Emergency Medicine Clinics* Jorge L. Falcon-Chevere,Jose Cabanas,2013-02-28 This issue focuses on Critical Skills and Procedures in the following topic areas Pediatric Orthopedics Vascular ENT Procedures

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Wilkinson Shaw,Edward John Mehren,1928 **School Education** ,1894 **Monthly Catalog of United States Government Publications** ,2002-07

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