HYDROCARBONS



Hydrocarbon Compounds

Hydrocarbon compounds are compounds that consist only of carbon and hydrogen atoms. Hydrocarbons can be divided into 2, namely:

a. Saturated Hydrocarbons: single bonds

including saturated hydrocarbons:

- Alkane: open chain (C_nH_{2n+2})
- a Cycloalkane : closed chain (C_nH_{≥n})

b. Unsaturated Hydrocarbons: double bonds

- Alkene : double bond (C_nH_{2n})
- Alkyne : triple bond (C_nH_{2n-2})
- Alkadiene : two double bonds (C_nH_{2n-2})



Nomenclature of Alkanes

The nomenclature of alkanes is the basis for the nomenclature of other carbon compounds. Therefore the nomenclature of alkanes must be thoroughly understood.

Here are the names of askanes from C1-C10:

Formula	Name	Formula	Name
CH ₄	methane	C6H14	hexane
C ₂ H ₆	ethane	C7H16	heptane
C ₃ H ₈	propane	C ₈ H ₁₈	octane
C4H10	butane	C9H20	nonane
C ₅ H ₁₂	pentane	C ₁₀ H ₂₂	decane

Atkene names :

 C_2H_4 = ethylene

 C_3H_6 = propene

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a must have for any practicing engineer or student in this area

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