

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

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Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 1. A column in the periodic table
- _____ 2. A row in the periodic table
- _____ 3. Group 1 elements
- _____ 4. Elements that are shiny and conduct electricity
- _____ 5. Group 17 elements

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 6. There are two representative elements of elements.
- _____ 7. Most elements are located in the periodic table are metals.
- _____ 8. Group 1A elements except for hydrogen are located in the alkali metals.
- _____ 9. Group 2A elements are the alkaline earth metals.
- _____ 10. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 11. Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ 12. Elements from groups of both metals and nonmetals are metalloids.

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Michael T. Madigan



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