

10.1 Periodic Trends



10.1 PERIODIC TRENDS FROM CONCEPTS

Section 10.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Hydrogen
_____	Helium
_____	Group 1 elements

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists as the periodic table.
- _____ 20. It has two the periodic table.
- _____ 21. Group 1 elements
- _____ 22. Elements that are strong oxidizing agents.
- _____ 23. Group 2 elements

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of elements.
- _____ 25. Most elements have two of the elements in the periodic table are metals.
- _____ 26. Group 1 elements except for hydrogen are metals in the solid state.
- _____ 27. Group 2 elements are the alkaline earth metals.
- _____ 28. Group 1 elements are highly reactive elements known as alkalis.
- _____ 29. Group 2 elements are very reactive elements known as alkaline earths.
- _____ 30. Alkaloids have properties of both metals and non-metals.

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R Pring



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Programmed Topics in General Chemistry Armine D. Paul, 1971 *Bulletin of the American Ceramic Society*, 1960
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