



States of Matter: Phase Changes

Phase	Description	Illustration
Evaporation	Liquid → _____ ❖ Water changes into a gas called _____ ❖ Heat energy is added ❖ Particles move faster and faster ❖ Examples include a _____ disappearing and clothes drying	
Condensation	Gas → _____ ❖ Water vapor turns into droplets of (liquid) _____ ❖ Heat energy is removed ❖ Particles _____ ❖ Examples include morning dew and a foggy mirror after a hot shower	
Freezing	Liquid → _____ ❖ Heat energy is _____ ❖ (Liquid) water turns into ice ❖ Particles slow down ❖ Examples include rain turning into _____ and juice freezing into a popsicle	
Melting	Solid → _____ ❖ Heat energy is _____ ❖ Ice turns into (liquid) water ❖ Particles move _____ ❖ Examples include a snowman thawing out and wax dripping down a candle	

❖ Matter can change from one phase to another with the addition or removal of _____

❖ Matter changing from one state to another is a _____

Study Guide Phase Changes

KJ Lindholm-Leary



Study Guide Phase Changes:

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