

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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Holt Chemistry Ralph Thomas Myers, 2004 **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24 **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 **Prentice Hall Chemistry**, 2000 *Te HS&T J* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02 **Holt Science and Technology** Holt Rinehart & Winston, 2004-02 **The School Science Review**, 2006 **Science Insights**, 1999 *Te HS&T 2007 Shrt Crs M* Holt Rinehart & Winston, 2007

Exploring Earth and Space Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter **The Arithmetic Teacher**, 1987 The Teaching Portfolio Peter Seldin, J. Elizabeth Miller, Clement A. Seldin, 2010-08-30 Praise for The Teaching Portfolio This new edition of a classic text has added invaluable immediately useful material It s a must read for faculty department chairs and academic administrators Irene W D Hecht director Department Leadership Programs American Council on Education This book offers a wealth of wisdom and materials It contains essential knowledge salient advice and an immediately useful model for faculty engaged in promotion or tenure Raymond L Calabrese professor of educational administration The Ohio State University The Teaching Portfolio provides the guidelines and models that faculty need to prepare quality portfolios plus the standards and practices required to evaluate them Linda B Nilson director Office of Teaching Effectiveness and Innovation Clemson University Focused on reflection sound assessment and collaboration this inspiring and practical book should be read by every graduate student faculty member and administrator John Zubizarreta professor of English Columbia College All the expanded and new sections of this book add real value but administrators and review committees will clearly benefit from the new section on how to evaluate portfolios with a validated template Barbara Hornum director Center for Academic Excellence Drexel University This book is practical insightful and immediately useful It s an essential resource for faculty seeking promotion tenure or who want to improve their teaching Michele Stocker Barkley faculty Department of Psychology Kishwaukee Community College The Teaching Portfolio has much to say to teachers of all ranks disciplines and institutions It offers a rich compendium of practical guidelines examples and resources Mary Deane Sorcinelli Associate Provost for Faculty Development University of Massachusetts Amherst Teaching portfolios help our Board on Rank and Tenure really understand the quality and value of individual teaching contributions Martha L Wharton Assistant Vice President for Academic Affairs and Diversity Loyola University Maryland *An Inquiry-Based Introduction to Engineering* Michelle Blum, 2022-09-20 The text introduces engineering to first year undergraduate students using Inquiry Based Learning IBL It draws on several different inquiry based instruction types such as confirmation inquiry structured inquiry guided inquiry and open inquiry and all of their common elements Professor Blum s approach emphasizes the student s role in the learning process empowering them in the classroom to explore the material ask questions and share ideas

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Emotionally Disturbed** National Information Center for Special Education Materials,1978 Illinois Chemistry Teacher
,1992 **School Library Journal** ,1986

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