

The Mole

Part 1: Mole and the Molecule

Answer the following questions using your knowledge of the mole.

- The atomic mass of carbon is 12.01. The mass of 1 mole of carbon is _____.
- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.
- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.
- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.
- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.
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- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.
- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.

Part 2: Molar Mass

- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.
- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.
- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.
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- The mass of 1 mole of carbon is 12.01 g. The mass of 1 mole of carbon is _____.

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What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g) What is the mass of 1 mole of carbon? (12.01 g)

2D

Given that 1 mole = 6.02 x 10²³ atoms and 1 mole of H is 1.01 grams

- How many atoms are in 1 mole?

1.406 x 10²⁴

- How many moles are in 1.01 grams of hydrogen (H)?

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Skill Sheet Chapter 1no 1

Ginny Stacey, Sally Fowler



Skill Sheet Chapter 1no 1:

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Resources in Education, 2001

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