

States of Matter

Review and Reinforce

1. The bonds of cohesiveness within the solid will be a liquid, and the liquid will be a gas.
2. The solid would lose the water drops because the particles in a solid are bound tightly together. The drops of the liquid will spread around and later evaporate because the particles in a liquid have spread out more. The drops of the liquid would continue to change and spread outwards because the particles in a gas move apart in many different directions.
3. Since the bonds of cohesiveness within the volume would stop the water. The volume of the liquid will eventually stop the water. The volume of the liquid would expand to fill the area of the larger container because the particles would spread apart.
4. Matter with a definite volume and a definite shape.
5. Matter with a definite volume but no shape will be a gas.
6. Matter with a definite shape or volume.
7. The molecules of a liquid are moving.
8. It refers to where the particles are not arranged in a regular pattern.
9. It refers to where the particles are arranged in a regular, repeating pattern.
10. A substance that flows.
11. The bonds of cohesiveness will prevent the molecules of a liquid from being further separated from together.

Changes of State

Review and Reinforce

1. Gas to liquid
2. Gas to liquid
3. Gas to liquid
4. Gas to liquid
5. Gas to liquid
6. Boiling point
7. condensation
8. evaporation
9. boiling
10. evaporation
11. melting
12. freezing
13. melting point
14. sublimation

Skills Reinforcement Activity 20 Answers

Kay P. Bland



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