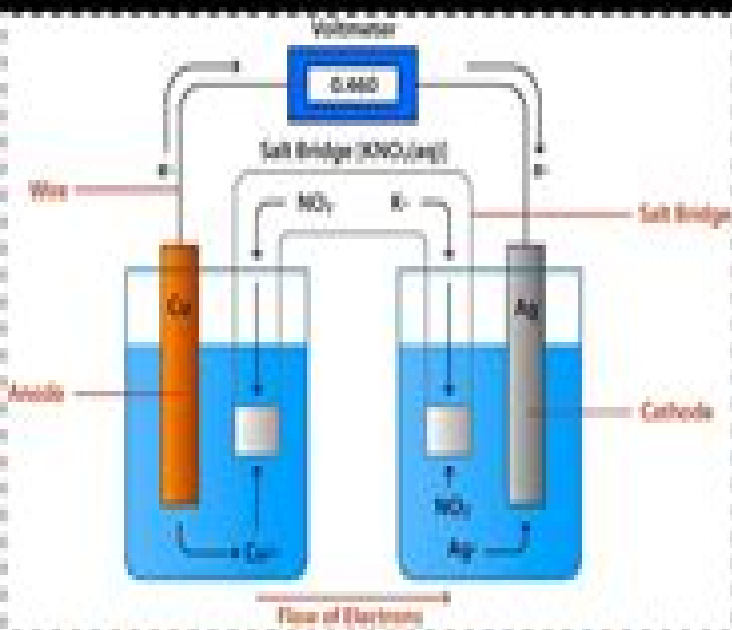
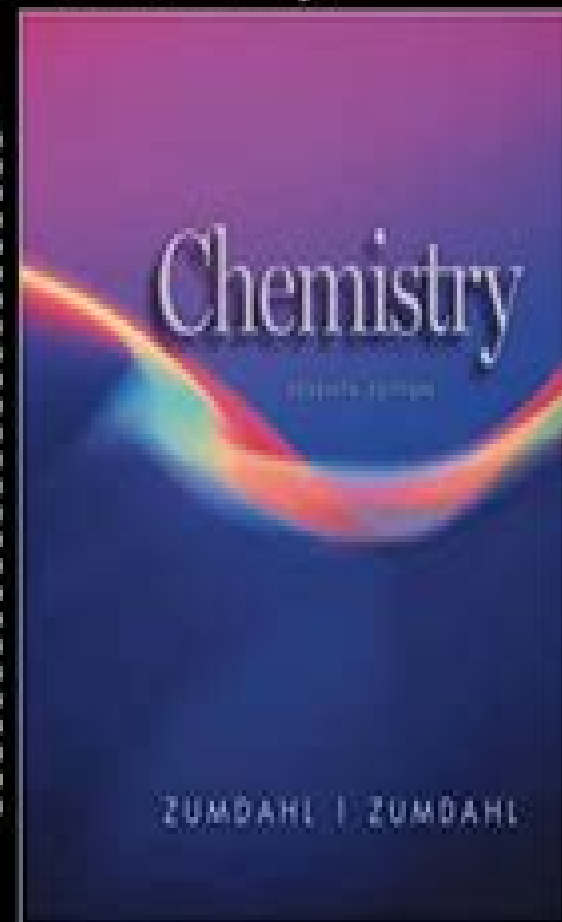


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## CHAPTER 17 (ELECTROCHEM)



Oxidizing Agent	Reducing Agent	Standard Potential (V)
F <sub>2</sub> + 2e <sup>-</sup> → 2F <sup>-</sup>	2F <sup>-</sup> → F <sub>2</sub> + 2e <sup>-</sup>	2.87
H <sub>2</sub> O <sub>2</sub> + 2H <sup>+</sup> + 2e <sup>-</sup> → 2H <sub>2</sub> O	2H <sub>2</sub> O → H <sub>2</sub> O <sub>2</sub> + 2H <sup>+</sup> + 2e <sup>-</sup>	1.78
MnO <sub>4</sub> <sup>-</sup> + 8H <sup>+</sup> + 5e <sup>-</sup> → Mn <sup>2+</sup> + 4H <sub>2</sub> O	Mn <sup>2+</sup> + 4H <sub>2</sub> O → MnO <sub>4</sub> <sup>-</sup> + 8H <sup>+</sup> + 5e <sup>-</sup>	1.51
Ag <sup>+</sup> + e <sup>-</sup> → Ag	Ag → Ag <sup>+</sup> + e <sup>-</sup>	1.50
Cl <sub>2</sub> + 2e <sup>-</sup> → 2Cl <sup>-</sup>	2Cl <sup>-</sup> → Cl <sub>2</sub> + 2e <sup>-</sup>	1.36
O <sub>2</sub> + 4H <sup>+</sup> + 4e <sup>-</sup> → 2H <sub>2</sub> O	2H <sub>2</sub> O → O <sub>2</sub> + 4H <sup>+</sup> + 4e <sup>-</sup>	1.23
O <sub>3</sub> /O <sub>2</sub> + 2H <sup>+</sup> + 2e <sup>-</sup> → O <sub>2</sub> + H <sub>2</sub> O	O <sub>2</sub> + H <sub>2</sub> O → O <sub>3</sub> /O <sub>2</sub> + 2H <sup>+</sup> + 2e <sup>-</sup>	1.23
Br <sub>2</sub> + 2e <sup>-</sup> → 2Br <sup>-</sup>	2Br <sup>-</sup> → Br <sub>2</sub> + 2e <sup>-</sup>	1.07
NO <sub>3</sub> <sup>-</sup> + 4H <sup>+</sup> + 3e <sup>-</sup> → NO + 2H <sub>2</sub> O	NO + 2H <sub>2</sub> O → NO <sub>3</sub> <sup>-</sup> + 4H <sup>+</sup> + 3e <sup>-</sup>	0.96
Ag <sup>+</sup> + e <sup>-</sup> → Ag	Ag → Ag <sup>+</sup> + e <sup>-</sup>	0.80
I <sub>2</sub> + 2e <sup>-</sup> → 2I <sup>-</sup>	2I <sup>-</sup> → I <sub>2</sub> + 2e <sup>-</sup>	0.54
Cu <sup>2+</sup> + e <sup>-</sup> → Cu <sup>+</sup>	Cu <sup>+</sup> → Cu <sup>2+</sup> + e <sup>-</sup>	0.52
O <sub>2</sub> + 2H <sub>2</sub> O + 4e <sup>-</sup> → 4OH <sup>-</sup>	4OH <sup>-</sup> → O <sub>2</sub> + 2H <sub>2</sub> O + 4e <sup>-</sup>	0.40
Cu <sup>2+</sup> + 2e <sup>-</sup> → Cu	Cu → Cu <sup>2+</sup> + 2e <sup>-</sup>	0.34
2H <sub>2</sub> O <sub>2</sub> + 2e <sup>-</sup> → 2H <sub>2</sub> O + O <sub>2</sub>	H <sub>2</sub> + O <sub>2</sub> → 2H <sub>2</sub> O <sub>2</sub> + 2e <sup>-</sup>	0.00
Hg <sup>2+</sup> + 2e <sup>-</sup> → Hg	Hg → Hg <sup>2+</sup> + 2e <sup>-</sup>	-0.85
Sn <sup>4+</sup> + 2e <sup>-</sup> → Sn <sup>2+</sup>	Sn <sup>2+</sup> → Sn <sup>4+</sup> + 2e <sup>-</sup>	-0.14
Fe <sup>3+</sup> + e <sup>-</sup> → Fe <sup>2+</sup>	Fe <sup>2+</sup> → Fe <sup>3+</sup> + e <sup>-</sup>	-0.77
Fe <sup>3+</sup> + e <sup>-</sup> → Fe <sup>2+</sup>	Fe <sup>2+</sup> → Fe <sup>3+</sup> + e <sup>-</sup>	-0.77
Fe <sup>3+</sup> + e <sup>-</sup> → Fe <sup>2+</sup>	Fe <sup>2+</sup> → Fe <sup>3+</sup> + e <sup>-</sup>	-0.77
Cr <sup>3+</sup> + 3e <sup>-</sup> → Cr	Cr → Cr <sup>3+</sup> + 3e <sup>-</sup>	-0.74
Zn <sup>2+</sup> + 2e <sup>-</sup> → Zn	Zn → Zn <sup>2+</sup> + 2e <sup>-</sup>	-0.76
2H <sub>2</sub> O + 2e <sup>-</sup> → H <sub>2</sub> + 2OH <sup>-</sup>	H <sub>2</sub> + 2OH <sup>-</sup> → 2H <sub>2</sub> O + 2e <sup>-</sup>	-0.83
Na <sup>+</sup> + e <sup>-</sup> → Na	Na → Na <sup>+</sup> + e <sup>-</sup>	-1.18
K <sup>+</sup> + e <sup>-</sup> → K	K → K <sup>+</sup> + e <sup>-</sup>	-1.80



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**Padhraic Smyth**



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## **Table of Contents Zumdahl Chemistry Chapter 17**

1. Understanding the eBook Zumdahl Chemistry Chapter 17
  - The Rise of Digital Reading Zumdahl Chemistry Chapter 17
  - Advantages of eBooks Over Traditional Books
2. Identifying Zumdahl Chemistry Chapter 17
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Zumdahl Chemistry Chapter 17
  - User-Friendly Interface
4. Exploring eBook Recommendations from Zumdahl Chemistry Chapter 17
  - Personalized Recommendations
  - Zumdahl Chemistry Chapter 17 User Reviews and Ratings
  - Zumdahl Chemistry Chapter 17 and Bestseller Lists
5. Accessing Zumdahl Chemistry Chapter 17 Free and Paid eBooks
  - Zumdahl Chemistry Chapter 17 Public Domain eBooks
  - Zumdahl Chemistry Chapter 17 eBook Subscription Services

- Zumdahl Chemistry Chapter 17 Budget-Friendly Options
- 6. Navigating Zumdahl Chemistry Chapter 17 eBook Formats
  - ePub, PDF, MOBI, and More
  - Zumdahl Chemistry Chapter 17 Compatibility with Devices
  - Zumdahl Chemistry Chapter 17 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Zumdahl Chemistry Chapter 17
  - Highlighting and Note-Taking Zumdahl Chemistry Chapter 17
  - Interactive Elements Zumdahl Chemistry Chapter 17
- 8. Staying Engaged with Zumdahl Chemistry Chapter 17
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Zumdahl Chemistry Chapter 17
- 9. Balancing eBooks and Physical Books Zumdahl Chemistry Chapter 17
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Zumdahl Chemistry Chapter 17
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Zumdahl Chemistry Chapter 17
  - Setting Reading Goals Zumdahl Chemistry Chapter 17
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Zumdahl Chemistry Chapter 17
  - Fact-Checking eBook Content of Zumdahl Chemistry Chapter 17
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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