

CHAPTER 6 REVIEW*Chemical Bonding***SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. a A chemical bond between atoms results from the attraction between the valence electrons and _____ of different atoms.
(a) nuclei (c) isotopes
(b) inner electrons (d) Lewis structures
2. b A covalent bond consists of
(a) a shared electron. (c) two different ions.
(b) a shared electron pair. (d) an octet of electrons.
3. a If two covalently bonded atoms are identical, the bond is identified as
(a) nonpolar covalent. (c) ionic.
(b) polar covalent. (d) dipolar.
4. b A covalent bond in which there is an unequal attraction for the shared electrons is
(a) nonpolar. (c) ionic.
(b) polar. (d) dipolar.
5. c Atoms with a strong attraction for electrons they share with another atom exhibit
(a) zero electronegativity. (c) high electronegativity.
(b) low electronegativity. (d) Lewis electronegativity.
6. c Bonds that possess between 5% and 50% ionic character are considered to be
(a) ionic. (c) polar covalent.
(b) pure covalent. (d) nonpolar covalent.
7. a The greater the electronegativity difference between two atoms bonded together, the greater the bond's percentage of
(a) ionic character. (c) metallic character.
(b) nonpolar character. (d) electron sharing.
8. The electrons involved in the formation of a chemical bond are called _____
 valence electrons
9. A chemical bond that results from the electrostatic attraction between positive and negative ions is called a(n) _____
 ionic bond

Toxins Unit Exam Answer Key Chem

Mr. Rohit Manglik

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the name bar.

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OECD Series on Testing and Assessment Guidance Document on Aquatic Toxicity Testing of Difficult Substances and Mixtures OECD,2019-05-16 This document provides the second edition of Guidance Document 23 on Aqueous Phase Aquatic Toxicity Testing of Difficult Test Chemicals Environmental Toxicology Mr. Rohit Manglik,2023-07-23 Toxic substances and environmental impact Includes toxicity exposure and health effects essential for environmental health and risk assessment studies **Resources in Education** ,1988 Serves as an index to Eric reports microform **Sittig's Handbook of Toxic and Hazardous Chemicals and Carcinogens** Richard P. Pohanish,2012 For more than a quarter century Sittig s Handbook of Toxic and Hazardous Chemicals and Carcinogens has proven to be among the most reliable easy to use and essential reference works on hazardous materials Sittig s 5th Edition remains the lone comprehensive work providing a vast array of critical information on the 2 100 most heavily used transported and regulated chemical substances of both occupational and environmental concern Information is the most vital resource anyone can have when dealing with potential hazardous substance accidents or acts of terror Sittig s provides extensive data for each of the 2 100 chemicals in a uniform format enabling fast and accurate decisions in any situation The chemicals are presented alphabetically and classified as a carcinogen hazardous substance hazardous waste or toxic pollutant This new edition contains extensively expanded information in all 28 fields for each chemical see table of contents and has been updated to keep pace with world events Chemicals classified as WMD have been included in the new edition as has more information frequently queried by first responders and frontline industrial safety personnel Sittig s Handbook is a globally recognized reference source providing full listings of the 2 000 most common hazardous chemicals making it the essential handbook for first line response to chemical spills and day to day chemical plant reference Entries have a full range of synonyms for each chemical including trade names to avoid confusion and enable quick and accurate location of the right information Authoritative and frequently updated Sittig provides a fully accurate source of information that engineers and emergency response services look to as a highly dependable reference both for emergencies and day to day engineering decisions

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