

THE FOLLOWING QUESTIONS ARE PRACTICE QUESTIONS FOR THE CSN CHEMISTRY PLACEMENT EXAM

The chemistry placement test is used to assess your preparation for General Chemistry I, CHEM 121 at CSN. The test is a standardized test that will examine your knowledge of chemistry and basic mathematical skills. You will be provided scratch paper, a Periodic Table of the Elements and a nonprogrammable scientific calculator. The following sample questions are intended only as a guide to show you the relative level of difficulty and example topics covered in the Chemistry Placement Exam. The actual exam will have a time limit of 45 minutes to answer 44 multiple choice questions. Click the following link to view a [Periodic Table](#).

MULTIPLE CHOICE. Choose the single choice that best answers the question.

- 1) How many total atoms are in the formula



A) 12 B) 14 C) 9 D) 8

- 2) How many electrons are in Br^- ?

A) 4 B) 7 C) 36 D) 34

- 3) What are the coefficients for the following reaction when it is properly balanced?



A) 2, 3, 6, 1 B) 6, 1, 3, 2
C) 2, 3, 1, 6 D) 2, 1, 1, 3

- 4) Given that $4 \text{ NH}_3 + 5 \text{ O}_2 \rightarrow 4 \text{ NO} + 6 \text{ H}_2\text{O}$, when 4.50 mol of H_2O are formed, the amount of NO formed is

A) 3.00 mol B) 6.75 mol
C) 1.50 mol D) 4.50 mol

- 5) What type of a reaction occurs when a sodium hydroxide solution is mixed with an acetic acid solution?

A) gas evolution
B) oxidation-reduction
C) acid-base neutralization
D) precipitation

- 6) Which of the following atoms has the smallest radius?

A) Ne B) O
C) Li D) B

- 7) What is the molecular geometry if you have a double bond, a single bond and 1 lone pair around the central atom?

A) trigonal pyramidal
B) linear
C) tetrahedral
D) bent

- 8) What volume of 12.0 M HCl is required to make 75.0 mL of 3.50 M HCl?

A) 0.560 mL B) 257 mL
C) 560. mL D) 21.9 mL

$$9) \frac{(2.80 \times 10^{-3})(1.28 \times 10^1)}{(3.20 \times 10^{-4})} = \underline{\hspace{2cm}}$$

A) 1.27×10^1 B) 2.0×10^2
C) 1.2×10^6 D) 1.12×10^2

- 10) Of the molecules below, only _____ is polar.

A) CH_4 B) SbF_5
C) CO_2 D) AsH_3

- 11) Which one of the following is a metalloid?

A) S B) Ge C) C D) Pb

- 12) A certain atom has a charge of -2 and has 16 protons and 16 neutrons in the nucleus. How many electrons does this atom possess?

A) 34 B) 14 C) 32 D) 18

- 13) How many moles of pyridine ($\text{C}_5\text{H}_5\text{N}$) are contained in 3.13 g of pyridine?

A) 0.00404 B) 25.3
C) 0.0396 D) 0.319

- 14) Solve the following equation for x:

$$4x - b = 2x + c$$

A) $c - b - 6$ B) $\frac{c + b}{2}$
C) $\frac{c + b}{6}$ D) $\frac{c - b}{2}$

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1. Understanding the eBook Toledo Chemistry Placement Exam Practice Test
 - The Rise of Digital Reading Toledo Chemistry Placement Exam Practice Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Toledo Chemistry Placement Exam Practice Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Toledo Chemistry Placement Exam Practice Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Toledo Chemistry Placement Exam Practice Test
 - Personalized Recommendations
 - Toledo Chemistry Placement Exam Practice Test User Reviews and Ratings
 - Toledo Chemistry Placement Exam Practice Test and Bestseller Lists
5. Accessing Toledo Chemistry Placement Exam Practice Test Free and Paid eBooks
 - Toledo Chemistry Placement Exam Practice Test Public Domain eBooks
 - Toledo Chemistry Placement Exam Practice Test eBook Subscription Services
 - Toledo Chemistry Placement Exam Practice Test Budget-Friendly Options

6. Navigating Toledo Chemistry Placement Exam Practice Test eBook Formats
 - ePub, PDF, MOBI, and More
 - Toledo Chemistry Placement Exam Practice Test Compatibility with Devices
 - Toledo Chemistry Placement Exam Practice Test Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Toledo Chemistry Placement Exam Practice Test
 - Highlighting and Note-Taking Toledo Chemistry Placement Exam Practice Test
 - Interactive Elements Toledo Chemistry Placement Exam Practice Test
8. Staying Engaged with Toledo Chemistry Placement Exam Practice Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Toledo Chemistry Placement Exam Practice Test
9. Balancing eBooks and Physical Books Toledo Chemistry Placement Exam Practice Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Toledo Chemistry Placement Exam Practice Test
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Toledo Chemistry Placement Exam Practice Test
 - Setting Reading Goals Toledo Chemistry Placement Exam Practice Test
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Toledo Chemistry Placement Exam Practice Test
 - Fact-Checking eBook Content of Toledo Chemistry Placement Exam Practice Test
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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