

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	58	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

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1. Understanding the eBook World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - The Rise of Digital Reading World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - User-Friendly Interface

4. Exploring eBook Recommendations from World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Personalized Recommendations
 - World Of Chemistry Chapter 3 Basic Review Worksheet Answers User Reviews and Ratings
 - World Of Chemistry Chapter 3 Basic Review Worksheet Answers and Bestseller Lists
5. Accessing World Of Chemistry Chapter 3 Basic Review Worksheet Answers Free and Paid eBooks
 - World Of Chemistry Chapter 3 Basic Review Worksheet Answers Public Domain eBooks
 - World Of Chemistry Chapter 3 Basic Review Worksheet Answers eBook Subscription Services
 - World Of Chemistry Chapter 3 Basic Review Worksheet Answers Budget-Friendly Options
6. Navigating World Of Chemistry Chapter 3 Basic Review Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - World Of Chemistry Chapter 3 Basic Review Worksheet Answers Compatibility with Devices
 - World Of Chemistry Chapter 3 Basic Review Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Highlighting and Note-Taking World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Interactive Elements World Of Chemistry Chapter 3 Basic Review Worksheet Answers
8. Staying Engaged with World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers World Of Chemistry Chapter 3 Basic Review Worksheet Answers
9. Balancing eBooks and Physical Books World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection World Of Chemistry Chapter 3 Basic Review Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Setting Reading Goals World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Fact-Checking eBook Content of World Of Chemistry Chapter 3 Basic Review Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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