

Wave Particle Duality Worksheet Answers

- 1) Explain, two phenomena of light, that provide evidence for the wave-particle duality of light.

The **photoelectric effect** and **diffraction** showcase that light behaves as a **particle** and a **wave** respectively. This phenomenon is known as **wave-particle duality**.

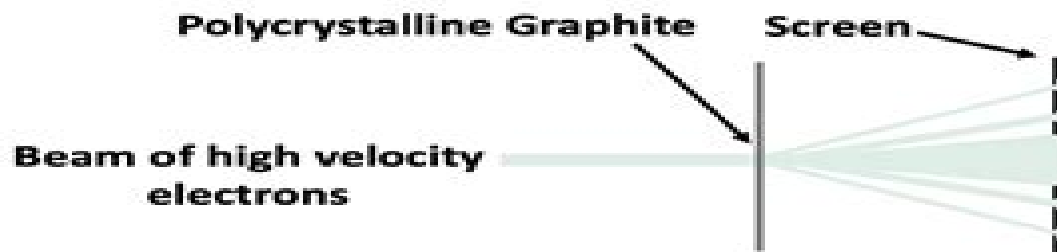
- 2) Write down de Broglie's equation, stating the units of each quantity. Explain how this equation can be modified.

$$\lambda = \frac{h}{p}$$

Wavelength (m) → λ Planck's constant ($6.63 \times 10^{-34} \text{ Js}$) → h Momentum (kgms^{-1}) → p

(Momentum = mass x velocity) $p = mv$. Therefore: $\lambda = \frac{h}{mv}$

- 3) Draw a labelled diagram illustrating electron diffraction. Describe what is observed and explain the condition necessary for this diffraction to take place.



A circular pattern of rings is observed. The **wavelength of electrons** must be **approximately equal to the spacing between the atoms** in the polycrystalline graphite. This is true and the value is around 10^{-10} m .

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