Net Ionic Equations Worksheet

Write balanced chemical equations for dissolving the following compounds in water:

Solid sodium chloride

Naclow -> Natow +Clicas)

Solid barium nitrate

Ba(NO3)2(a0) -> Ba2(ac) + 2NO2-(a0)

Solid magnesium hydroxide

Mg(OH)2(0) -> Ma2+(00) + 20H (00)

Gaseous hydrochloric acid

HClas + Class

Solid calcium phosphate

Cax(PO+)=163-3CATOON+2PO+3-

Solid iron (III) sulfide

Fe2S3(8) -> 2 Fe+3(00)+35-2(00)

7. Gaseous carbon dioxide (Hint: Carbon dioxide will form carbonic acid first then decomposes into ions) CO2(a) -> Ha CO2/an -> 2Htran + CO2 (00)

Write balanced chemical equations and then net ionic equations for the following reactions:

8. Aqueous sodium sulfate and aqueous barium chloride

1Nazsonan+1Baclz->2Naclan+1Bason(s) 2 Nature + SO4 2000+ Bas+(00 +261 cm) -> 2 Nature + 261 cms+ BasO4(0) 9. Aqueous lithium bromide and aqueous silver nitrate

LiBras + AgNOscae) -> LiNOscae) + AgBras

10. Aqueous lead II acetate and aqueous sodium carbonate

Pb(C2H3O2)2(00)+ NazCO3(00)-> PbCO3(5)+2 NaC2H3O2(00) Pb+2(02)+ 2024302 (00)+ 2010+(00)+ CO3-2(02)-> PbCO3(5)+2Noting +2624302(00)

Pb+2 (as) + CO3 (ag) -> PbCO3(S)

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Holt Rinehart & Winston

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