C	UIZIZZ	N.	AME:
Intermolecular Forces 19 Questions		CLASS: DATE:	
A,	Strong dipole-dipole bonds between water molecules	В	Dispersion forces which are present in all molecules
	Strong hydrogen bonds between water molecules	D	Asymmetrical shape of the polar bonds
2.	Does HCI have hydrogen bonding?		
A.	no	8	yes
3.	Does CH ₄ have hydrogen bonding?		
Δų	yes	В	no
4	All molecules have London forces betwee bonding are so much stronger that whe forces. Which of these has ONLY London	n they	are present we can ignore London
A,	12	В	OCl ₂
	NH ₃	D	SH ₂
5.	Which substance would have the weake:	st inte	rmolecular forces of attraction?
A. T	CH ₄	[B]	H ₂ O

° н—й—н

Intermolecular forces for: NH₃

D

A Dipole dipole

NaCI

B Dispersion Force

 MgF_2

States Of Matter And Intermolecular Forces Test

Dr. S C Rastogi, ,Er. Meera Goyal

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