

Section A: Principles of Chemistry

A1 States of Matter

No.	Answers	Further explanations
1	C	
2	D	$\text{NH}_3(\text{g}) + \text{HCl}(\text{g}) \longrightarrow \text{NH}_4\text{Cl}(\text{s})$ <p>The concentrated ammonia solution gives off ammonia gas and the concentrated hydrochloric acid gives off hydrogen chloride gas, shown by the state symbols (g). The particles of both gases diffuse along the tube, collide and react to form a white ring composed of solid white ammonium chloride particles, shown by the state symbol (s).</p>
3	A	<p>Ammonia molecules diffuse faster than hydrogen chloride molecules. Ammonia molecules are smaller and lighter than hydrogen chloride molecules, so they diffuse faster along the tube, and the particles of the two gases collide and react closer to the source of the hydrogen chloride.</p>
4	C	
5	A	<p>Increased in length and become rigid</p> <p>The paw-paw strip is composed of living cells. Each cell is surrounded by a differentially permeable membrane and contains cytoplasm containing about 80% water. By osmosis, water enters each cell causing it to swell slightly. This causes the strip to increase in length and become rigid.</p>
6	B	<p>Moved down</p> <p>The 20% sucrose solution inside the thistle funnel contains more water than the 40% sucrose solution in the beaker. By osmosis, water moves out of the solution in the thistle funnel through the differentially permeable membrane into the solution in the beaker. This decreases the volume of the solution in the thistle funnel which causes the meniscus to move down.</p>

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Neil Hooley



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Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation *Pearson Edexcel A Level Chemistry (Year 1 and Year 2)* Andrew Hunt, Graham Curtis, Graham Hill, 2019-07-15 Develop and assess your students knowledge and skills throughout A level with worked examples practical assessment guidance and differentiated end of topic questions in this updated all in one textbook for Years 1 and 2 Combining everything your students need to know for the Pearson Edexcel A level Chemistry specification this revised textbook will Identify the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Student Book Provide support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Improve mathematical skills with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offer plenty of practice with Test yourself questions to help students assess their understanding and measure progress Encourage further reading and study with short passages of extension material Develop understanding with free online access to Test yourself answers and an extended glossary **Longman Science Chemistry 10** Kohli Nitin, 2008-09 Longman Science Chemistry 9 Kohli Nitin, 2008-09 *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science*, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Sif Chemistry NI Tb Rex M. Heyworth, 2007 *Matter*, 2004 Everything in the universe regardless of its size shape color or physical state is made up of matter *The Student and Intellectual Observer of Science, Literature and Art*, 1868 **Job interview questions and answers for employment on Offshore Drilling Platforms** PETROGAV INTERNATIONAL, 2020-06-28 The job interview is probably the most important step you will take in your job search journey Because it s always important to be prepared to respond effectively to the questions that employers typically ask at a job

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Young Scientist Series ICSE Chemistry 7 Lorraine Mirza, Krishnakali Gupta,

Working Mother, 2002-10 The magazine that helps career moms balance their personal and professional lives

The Chemical News and Journal of Industrial Science, 1917 **Campbell Biology Australian and New Zealand Edition** Jane B. Reece, Noel Meyers, Lisa A. Urry, Michael L. Cain, Steven A.

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