

NAME: _____

LAST SECTION # _____

INSTRUCTIONS: PLEASE READ & GOOD LUCK !!! (10 min)

- Put your name and identification # on page 1 of test.
- Exam time is 1:00 to 1:45:00 pm. After test ends, keep your seat and see those.
- Be generous for questions, but don't leave any unanswered.
- **DO NOT TEAR OFF THE FIRST PAGE OF THE EXAM**
- Leave all pages of your exam stapled together. (You may separate for test pages.)
- On problems, show all your work for full credit (partial credit). No credit for just equations.
- Credit given ONLY for problems worked on first, not being circumvented using paper.
- On multiple problems, "double (generally)" credit will not exceed any limit of problems.
- When you finish, turn off Periodic Table & put exam in envelope with your questions.

POSSIBLY USEFUL INFORMATION:

$$R = 0.08206 \frac{\text{L atm}}{\text{mol K}} = 8.314 \frac{\text{J}}{\text{mol K}}$$

$$\text{Density of water} = 1.00 \frac{\text{g}}{\text{mL}}$$

NOMENCLATURE: (2 pts each. Mark) Fill in blank with appropriate answer.SYNOPSISNAME

1. _____
2. _____
3. _____
4. _____
5. _____ Address information _____

MULTIPLE CHOICE: (2 pts each question)6. _____ Which of the following is CORRECT for Na_2SO_4 aqueous solution?

- a. 0.000 M b. 0.000 M c. 0.000 M d. 0.000 M e. 0.000 M

Wingrave Lab Manual

Padhraic Smyth



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