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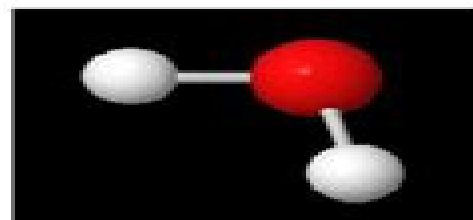
Student Exploration: Molecule Builder

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical bond, chemical formula, covalent bond, double covalent bond, cyclic molecule, hydrocarbon, ionic bond, isomers, Lewis structure, molecule, organic compound

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Look at the water **molecule**. In the molecule, **chemical bonds** are shown as lines between atoms.



1. The red atom in the middle is oxygen. How many bonds does oxygen have?

Two covalent bonds.

2. The two white atoms are hydrogens. How many bonds does one hydrogen atom have?

One

3. The **chemical formula** of water is H_2O . In this formula, what do you think the "2" between the H and O refers to?

Two hydrogen atoms

Gizmo Warm-up

In the Molecule Builder Gizmo, drag an oxygen atom into the white simulation area. The short blue lines extending from the atom represent possible chemical bonds.



1. How many bonds can oxygen form?

two

2. Drag a hydrogen atom into the area. How many bonds can oxygen form?

one

3. Drag the hydrogen atom on top of the oxygen atom. What happens?

They bond together.

The black line represents a chemical bond between the hydrogen and oxygen atoms.

4. Bond another hydrogen atom to the other side of the oxygen atom. Now, enter the formula "H₂O" into the **Type the formula below** box. Congratulations! You have created a water

Virtual Lab Exploration Guide

CH Cherryholmes



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