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COURSE

Vista 2programming Guide

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen

ion from an acid acidic substances are usually identified by their sour taste

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other s chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is ce acid base water salt nonumber

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web the agenda for the superior greenstone dsb 2022 08 regular board meeting august 29 2022 be accepted and approved as amended carried 4 0 disclosures of interest re open session there were no disclosures of interest offered at this time that pertained to the agenda 5 0 minutes board meetings and board committee meetings 5 1 board

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