

Electron Configuration Practice
Chemistry

Name: _____

Due Date: _____

How to write an electron configuration:

- Determine the total number of electrons to be represented.
- Use the Aufbau principle to fill the orbitals with electrons for elements 1-23. Refer to electron configuration periodic table for elements after 23
- The sum of the superscripts should equal the total number of electrons. Example:
 $12\text{Mg } 1s^2 2s^2 2p^6 3s^2$

I. Configuration Writing Practice

- Write a ground state electron configuration for each neutral atom. Ground state means that all of the lowest possible energy levels (up to the proper number of electrons for the element) are filled.

1. Na: $1s^2 2s^2 2p^6 3s^1$

2. Pb: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^2$

3. Sr: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2$

4. U: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^6 7s^2 5f^3 6d^1$

5. N: $1s^2 2s^2 2p^3$

* 6. Ag: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1 4d^{10}$

7. Ti: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^2$

8. Ce: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^1 5d^1$

9. Cl: $1s^2 2s^2 2p^6 3s^2 3p^5$

10. Hg: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10}$

Skill Practice 13 Configuration Practice Answers

Wendell Odom

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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Human Resource Development Jeff Gold, Rick

Holden,Paul Iles,2017-09-16 This core textbook edited by five leading scholars of the subject provides a comprehensive overview of the key topics debates and themes in this increasingly important field Balancing research led theory with industry best practice to provide students with a definitive overview of HRD the book draws on the international experience of its authors to tackle topics as diverse as leadership and managing development change and diversity workplace learning and graduate employability The book s approachable yet thorough writing style and lively presentation helps students to understand the topic from a critical perspective while also demonstrating how HRD plays out in reality This is an essential textbook for undergraduate postgraduate and MBA students of Human Resource Development on HRD or Business and Management degree programmes New to this Edition New contributors and revised content including additional coverage of careers career management and employability More international coverage especially of the EU Inclusion of topical subjects including employee engagement skills shortage and business partnering Improved student friendly pedagogy and updated figures and diagrams to appeal to different learning styles Thoroughly updated references and web links [Microsoft Windows Server 2003 ,2004](#)

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Serious Games Stefan Göbel,Minhua Ma,Jannicke Baalsrud Hauge,Manuel

Fradinho Oliveira,Josef Wiemeyer,Viktor Wendel,2015-05-27 This book constitutes the proceedings of the First Joint International Conference on Serious Games JCSG 2015 held in Huddersfield UK in June 2015 This conference bundles the activities of the International Conference on Serious Games Development and Applications SGDA and the Conference on Serious Games GameDays The total of 12 full papers and 5 short papers was carefully reviewed and selected from 31 submissions The book also contains one full invited talk The papers were organized in topical sections named games for health games for learning games for other purposes game design and development and poster and demo papers

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