

Concept Review Section 8.2

How Substances Dissolve

1. Explain how you can speed up the dissolving process when preparing a hot saturated solution.

2. Explain why water is sometimes referred to as the universal solvent.

3. Describe a solid molecule. Explain how the forces between molecules would be affected if the molecule had individual charges.

4. The water is not dissolving the ionic solid. Change, stir, add, and separate. How do you affect the rate of dissolution? Explain the three components:
 a. identify the particular species involved in the stirring

- b. Explain the strength of the particular forces between molecules and how much of energy and distance were involved and a distance of effect is.

- c. How else will happen when water molecules interact with an ionic solid?

Skills Worksheet Concept Review Section How Substances Dissolve Answers

Yijin Wang



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

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