

## Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON\_METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.)  $H_2$  hydrogen is diatomic:



2.)  $F_2$  fluorine is diatomic:



3.)  $O_2$  oxygen is diatomic:



4.)  $N_2$  nitrogen is diatomic. Is this a triple bond?



5.)  $BF_3$  you need 3 fluorine atoms here:



6.) Ammonia  $NH_3$  hint: how many hydrogen atoms are needed?



7.) Carbon dioxide  $CO_2$



8.) Methane  $CH_4$  careful here 4 hydrogen atoms needed



9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!



10.)  $SO_2$  hint: one pair of electrons from sulfur must be slit up for this one to work.



# The Nature Of Covalent Bonding Answers

**Mr. Ajeet Singh**



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Delve into the emotional tapestry woven by in Dive into the Emotion of **The Nature Of Covalent Bonding Answers** . This ebook, available for download in a PDF format ( PDF Size: \*), is more than just words on a page; itis a journey of connection and profound emotion. Immerse yourself in narratives that tug at your heartstrings. Download now to experience the pulse of each page and let your emotions run wild.

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