Dr. - Heder Mass

- 70. Come Ser the following:
 - a. If the formula of a compound is known (e.g. CytyCl), and you find the masses of the component atoms from the Periodic Table IC = 12 011g, H = 1 0079, & Cl = 35.455g). HOW would you find the molar mass of the compound?
 - b. Calculate the molar mass of one mole of CJNJCI.
 - c. Calculate the mass of 5 miles of CJHJCI

E - Percent Composition of Compounds

- Provide the word equation describing how you solve the mass percent for a given element in a compound.
- 9) Fernicillin is an important antibiotic that has the chemical formula C_wH_wN_ySO_w. Calculate the mass
- percent of EACH element in this compound.

F - Empirical Formulas

- 10) Consider the following:
- Taking in to account the formula for glucose, C_aH_{ay}O_a, what information can be gethered about this compound?
 - b. What ratio of CH O is present when considering the formula above?
 - c. Now take in to account the following Formulas: CyHyOu, CyHyOu, and CyHyOu. What do these formulas have in common with glucose?
- 11) Using the word "ratio" in the descriptions, explain what the difference is between empirical and

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LL Leslie

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