

Date Recd.: _____ Page No.: _____

Subject: Chemistry

The Mole

Quest 1. Fill in the blanks.

Write the correct chemical formulae in the space provided.

- The atomic number of sodium is 11, the chemical symbol of sodium is Na.
- The atomic number of oxygen is 8, the chemical symbol of oxygen is O.
- The molecular mass of H_2O is 18 and the chemical formula of water is H_2O .
- The molecular mass of CO_2 is 44 and the chemical formula of carbon dioxide is CO_2 .
- The molecular mass of CH_4 is 16 and the chemical formula of methane is CH_4 .

Quest 2. Answer questions.

- How many atoms are there in 1 mole of hydrogen? 6.023 $\times 10^{23}$ atoms.

or

 6.023×10^{23}

How many atoms are there in 1 mole of oxygen? 12.046 $\times 10^{23}$ atoms.

 12.046×10^{23}

Given that 1 mole = 6.023×10^{23} atoms and 1 mole of H_2 = 2.016 grams

- How many atoms are in 1 mole?

 1.2046×10^{24}

- How many moles are in 2.016 grams of hydrogen (H_2)?

 1

Skill Sheet Chapter 1no 1

Rachel S Tattersall



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