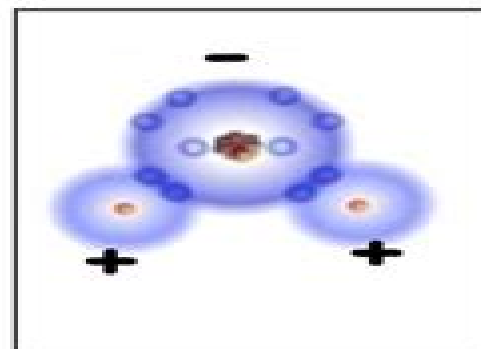
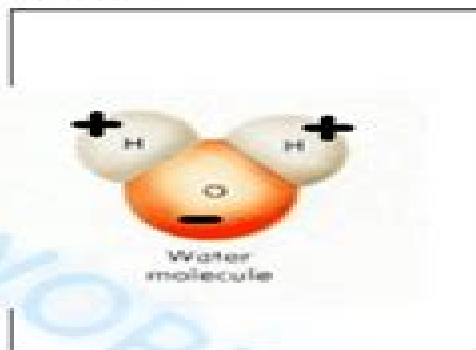
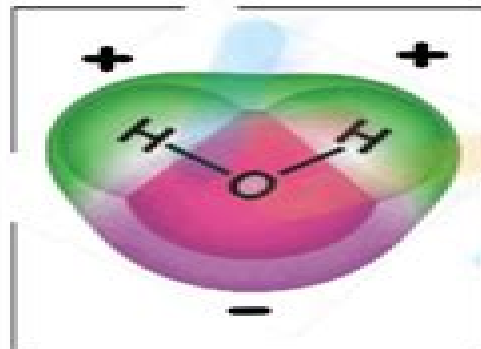


Properties of Water Activity

Name _____ Class _____

Water is everywhere. It's in the air we breathe. It's in our sink faucets. It's in every cell of our body. Water is an unusual substance with **many special properties**.

INTRODUCTION: The Water Molecule



What do you observe about the water molecule in these models?

1. _____
2. _____
3. _____

READ: Water is a polar molecule. That is, while a water molecule is neutral as a whole, it has a slightly positive charge on the oxygen side and a slightly negative charge on the hydrogen side. Water is found on Earth as a solid, liquid, and gas. This is because Earth is a very special planet with just the right range of temperatures and air pressures. Earth is said to be at the **triple point** for water.

QUESTIONS:

1. How many atoms are in a water molecule? Be specific.
2. Why does a water molecule have **polarity**?
3. Why is Earth said to be at the **triple point** for water?

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AN Whitehead



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Table of Contents Skills Worksheet Section Properties Of Ocean Water

1. Understanding the eBook Skills Worksheet Section Properties Of Ocean Water
 - The Rise of Digital Reading Skills Worksheet Section Properties Of Ocean Water
 - Advantages of eBooks Over Traditional Books
2. Identifying Skills Worksheet Section Properties Of Ocean Water
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Skills Worksheet Section Properties Of Ocean Water
 - User-Friendly Interface
4. Exploring eBook Recommendations from Skills Worksheet Section Properties Of Ocean Water

- Personalized Recommendations
- Skills Worksheet Section Properties Of Ocean Water User Reviews and Ratings
- Skills Worksheet Section Properties Of Ocean Water and Bestseller Lists
- 5. Accessing Skills Worksheet Section Properties Of Ocean Water Free and Paid eBooks
 - Skills Worksheet Section Properties Of Ocean Water Public Domain eBooks
 - Skills Worksheet Section Properties Of Ocean Water eBook Subscription Services
 - Skills Worksheet Section Properties Of Ocean Water Budget-Friendly Options
- 6. Navigating Skills Worksheet Section Properties Of Ocean Water eBook Formats
 - ePub, PDF, MOBI, and More
 - Skills Worksheet Section Properties Of Ocean Water Compatibility with Devices
 - Skills Worksheet Section Properties Of Ocean Water Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Skills Worksheet Section Properties Of Ocean Water
 - Highlighting and Note-Taking Skills Worksheet Section Properties Of Ocean Water
 - Interactive Elements Skills Worksheet Section Properties Of Ocean Water
- 8. Staying Engaged with Skills Worksheet Section Properties Of Ocean Water
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Skills Worksheet Section Properties Of Ocean Water
- 9. Balancing eBooks and Physical Books Skills Worksheet Section Properties Of Ocean Water
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Skills Worksheet Section Properties Of Ocean Water
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Skills Worksheet Section Properties Of Ocean Water
 - Setting Reading Goals Skills Worksheet Section Properties Of Ocean Water
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Skills Worksheet Section Properties Of Ocean Water

- Fact-Checking eBook Content of Skills Worksheet Section Properties Of Ocean Water
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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