

Chemistry: Atom Unit Study Guide

24. Fill in the boxes below for the following concepts:

1. Isotope
2. Atomic Number
3. Mass Number
4. Proton
5. Neutron
6. Electron

25. Draw a Bohr model of the element

26. Complete the following table:

27. Draw a Bohr model of the element

28. Using the periodic table, determine the number of protons, neutrons, and electrons of ^{13}C .

Name: _____ Period: _____

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Matching: Draw a line to match the words listed to the definitions.

- | | |
|------------------|--|
| 1. Isotope | — Atoms of the same element with different numbers of neutrons. |
| 2. Atomic Number | — The subscript particle has a positive charge. |
| 3. Mass Number | — The subscript particle has a negative charge. |
| 4. Proton | — The subscript particle has no charge (neutral). |
| 5. Neutron | — The number indicates the total number of protons and neutrons in an atom. |
| 6. Electron | — This number indicates the number of protons in an atom and defines an element. |

Circle the Blank.

Word Bank: Isotope, Neutrons, Empty Space, Neutrons, Space, Protons, Number
(Words can be used more than once)

- Where is most of an atom's mass found? _____
- An atom is mostly _____.
- Which subatomic particle is the atomic number equal to? _____
- The atomic number is equal to _____, _____, the number indicates the amount of _____ in _____.
- Isotopes of the same atom have the same _____ number but different _____ numbers.

Calculate.

- Determine the number of protons, neutrons, and electrons of ^{13}C .
- How many neutrons does the element Silver (Ag) have?

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B Lingard



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Organic Chemistry Study Guide Robert J. Ouellette, J. David Rawn, 2014-11-04 *Organic Chemistry Study Guide: Key Concepts, Problems, and Solutions* features hundreds of problems from the companion book *Organic Chemistry* and includes solutions for every problem. Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject. Organic chemistry is a constantly evolving field that has great relevance for all scientists, not just chemists. For chemical engineers, understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant. For biologists and health

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