

ANSWER KEY

ATOMS AND ELECTRONS STUDY GUIDE

Learning Goals: (a) (a.i) P. 4.1 - Explain a new scientific theory of atomic-scale phenomena (atomic theory) by describing changes in the atomic model over time, including those changes that are consistent with experimental evidence. (a.i.ii) P. 4.2 - Explain the scientific theory of atomic-scale phenomena by describing the scientific changes in terms of changing scientific understandings, including scientific theory that are not fully accepted by some of their peers, including changes and limitations within the team. (a.i.iii) P. 4.3 - Explain properties of atoms and their patterns in the periodic table in terms of their composition of their components, related to P. 4.4, as changes in the scientific understandings of their first fundamental principles (scientific theory, experimental evidence, model building, theory building).

ANSWER KEY

Directions: Read the atomic theory selected on the correct understanding.

ANSWER	ANSWER KEY	ANSWER KEY: SELECTED UNDERSTANDING
1C	Continuous (a.i.ii) P. 4.1	A. Everything is made of atoms (continuous) (a.i.ii) P. 4.2
2A	Discrete (a.i.ii) P. 4.1	B. Matter is made of atoms (discrete) (a.i.ii) P. 4.2
3B	Continuous (a.i.ii) P. 4.1	C. Matter is made of atoms (continuous) (a.i.ii) P. 4.2
4D	Discrete (a.i.ii) P. 4.1	D. Matter is made of atoms (discrete) (a.i.ii) P. 4.2
5C	Discrete (a.i.ii) P. 4.1	E. Matter is made of atoms (discrete) (a.i.ii) P. 4.2
6B	Discrete (a.i.ii) P. 4.1	F. Matter is made of atoms (discrete) (a.i.ii) P. 4.2

Directions: Use the periodic table to answer the following questions:

- 1C** - P. 4.1.ii - The number of protons equals the number of electrons.
- atomic mass** minus the atomic number will equal the **number of neutrons**.
- P. 4.1.ii** - **1C** - protons are positive (positively charged) and neutrons are neutral (neutral).

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Minjie Lin



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