

1. Write your answer.

Math Skills

Determining pH

Below are study-unit sample problems and solutions, where sample-problem problems are in separate boxes of paper. Write your answers to the sample-problem.

Problem 1a

Below is the pH of a 0.1 M solution of the strong acid HCl dissolved in water?

Solution

Step 1: Write the given and unknown values.

Given: concentration of HCl in solution is 0.100 M

Unknown: pH

Step 2: Consider the strong electrolyte of hydrochloric acid.

concentration of HCl in solution is 0.100 M

HCl is a strong acid, so the concentration of hydrogen ions in the solution is equal to the concentration of HCl.

concentration of H⁺ ions in concentration 0.100 M is 0.100 M

Step 3: Calculate the H⁺ concentration pH

pH is the negative of the power of ten used to describe the concentration of H⁺ ions

concentration of H⁺ ions is 0.100 M

pH = -log(0.100)

Problem 1b

1. A strong acid, HCl, has been dissolved into a known volume of the solution. Below is the concentration of HCl, what is the pH of the solution?

2. A weak acid, HCl, is dissolved into a known volume of the solution. Below is the concentration of HCl, what is the pH of the solution?

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