

Math Skills

Specific Heat

How much energy must be transferred as heat to 200 kg of water in a bathtub to raise the water's temperature from 25 °C to 37 °C?

1. List the given and the unknown values.

Given: $\Delta T = 37\text{ °C} - 25\text{ °C} = 12\text{ °C} = 12\text{ K}$

$$\Delta T = 12\text{ K}$$

$$m = 200\text{ kg}$$

$$c = 4,186\text{ J/kg}\cdot\text{K}$$

Unknown: *energy* = ? J

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Wenbin Ji



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