

KEY

19. Consider the rate of the following reaction:



- a) Is it dependent on temperature? yes. Explain your answer.

Mostly all reactions are temperature dependent.

- b) Is it dependent on pressure? no. Explain your answer.

No gaseous reactants

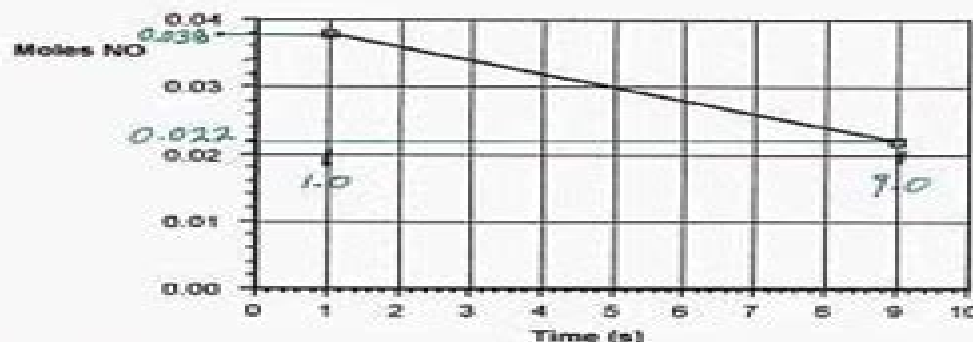
- c) Is it dependent on surface area? yes. Explain your answer.

Heterogeneous reactions. (s) and (aq) reactants

20. Consider the following reaction:



Data collected for the above reaction was used to construct the following graph:



From this graph, determine the rate of reaction in moles of NO consumed per second.

$$\text{slope} = - \frac{(0.038 - 0.022) \text{ mol}}{(9.0 - 1.0) \text{ s}} = \frac{0.016 \text{ mol}}{8 \text{ s}} = -0.002 \text{ mol NO/s}$$

$$\text{so rate} = 0.002 \text{ mol NO/s}$$

Answer 0.002 mol NO/s

Topic 8 Kinetics And Equilibrium Answer Key

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Topic 8 Kinetics And Equilibrium Answer Key:

Reviewing **Topic 8 Kinetics And Equilibrium Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Topic 8 Kinetics And Equilibrium Answer Key**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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