

THE ULTIMATE GENERAL CHEMISTRY 1 STUDY GUIDE

❖ Matter and Measurement

- Matter: anything that has mass and occupies space
- Properties of matter: physical properties and chemical properties
- States of matter: solid, liquid, gas
- Measurements: SI Units, significant figures, dimensional analysis

❖ Atoms, Molecules, and Ions

- Atomic structure: protons, neutrons, and electrons
- Periodic table and trends: electronegativity, ionization energy, electron affinity
- Chemical bonding: ionic bonding, covalent bonding, metallic bonding
- Chemical formulas and equations: empirical and molecular formulas, balancing chemical equations

❖ Stoichiometry

- Limiting reactants and percent yield
- Mole concept and Avogadro's number
- Molarity and molality
- Gas laws: Boyle's law, Charles's law, Gay-Lussac's law, Avogadro's law, Ideal gas law

❖ Chemical Reactions and Equations

- Types of chemical reactions: synthesis, decomposition, single replacement, double replacement, combustion
- Redox reactions and balancing redox equations
- Reaction rates: factors affecting reaction rates, rate laws, and reaction mechanisms

❖ Gases

- Gas laws: Boyle's law, Charles's law, Gay-Lussac's law, Avogadro's law, Ideal gas law
- Kinetic Molecular Theory of gases: pressure, temperature, volume, and the number of moles of gas particles
- Natural gases: van der Waals equation

Topic Eight Study Guide Chemistry

Donovan Young



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their examination at anytime from anywhere This book highly useful for those candidates who willing to clear the exams with the good grades and planning to study from the great institutes

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