



# Toshiba Dvd Cassette Recorder Dvr620 Manual

**Samsung Electronics**



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volume of gas moles  $V_m$  rearranging this moles of gas  $\frac{V}{V_m}$

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web jun 8 2023  $V = 1.00 \text{ mol} \times 0.08206 \text{ L atm mol}^{-1} \text{ K}^{-1} \times 273 \text{ K} = 22.4 \text{ L}$  in other words 1 mole of a gas will occupy 22.4 L at STP  
assuming ideal gas behavior at STP the volume of a gas is only dependent on number of moles of

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web aug 3 2022 volumes of gases and liquids are measured in cubic centimetres  $\text{cm}^3$  or cubic decimetres  $\text{dm}^3$  or litres L 1  
mole of any gas at room temperature and pressure occupies  $24 \text{ dm}^3$   $24\,000 \text{ cm}^3$  this value is known as the molar volume

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