Check It (37 to 39)

1. C 2. B 3. D 4. A 5. B 6. A 7. D 8. B 9. B

10a) calcium oxide and potassium chloride

- b) carbon dioxide and methane
- c) calcium oxide or potassium chloride (any one, an ionic one)
- d) carbon dioxide or methane (any one, a covalent one)
- e) helium (a noble gas, is monatomic, as single atoms)
- 11a) C:(2,4); O:(2,6); Na:(2,8,1); Si:(2,8,4); Cl:(2,8,7)
- b)(i) No. of neutrons = Nucleon no no of protons = 9 4 = 5
 - (ii) No. of neutrons = 28 14 = 14
- Beryllium and sodium (are metals, but, diamond does not conduct)
- d)(i) NaCl, sodium chloride
 - (ii) Na₂O, sodium oxide
 - (iii) CO₂, carbon dioxide or CO, carbon monoxide
 - (iv) Na₂CO₃, sodium carbonate (C and O forms the radical ion CO₃²⁻)
- e) Carbon dioxide is a covalent compound with a <u>simple</u>
 <u>molecular structure</u>. It is made up of small molecules with
 <u>weak intermolecular forces of attraction</u> between
 molecules. <u>Little amount of heat energy</u> is needed to
 overcome these weak forces. Thus, it has a <u>low boiling</u>
 point and is a gas at r.t.p.

Notes: I. Type of Bonding - Ionic & Covalent

Topic 6 Bonding Prentice Hall Textbook Answer

DJ Losen

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