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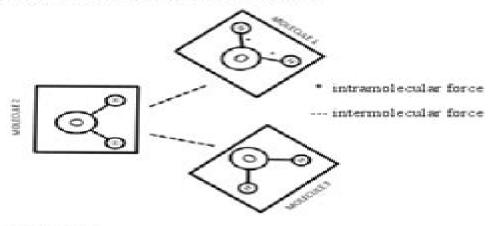
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POGIL: Intermolecular Forces

Model 1: What is an intermolecular force?

As you have learned matter is made up of discrete particles called atoms, which chemically combine to form molecules. Molecules do not exist as independent units, in fact, groups of molecules "stick together" in order to form liquids and solids. The forces that hold groups of molecules together are intermolecular forces. Without intermolecular forces, the world as we know it would not be the same.

Figure 1: Intramolecular and Intermolecular Forces



Critical Thinking Questions:

- What specific molecule is represented inside each box in Figure 1?
- In relation to the box for molecule 1, where do the intramolecular forces exist in Figure 1 – inside the boxes or outside the boxes?
- Based on the intramolecular forces for molecule 1, draw similar asterisks (*) for the intramolecular forces on the diagram for molecules 2 and 3.
- 4. In relation to the molecule, where do intramolecular forces tend to occur within the molecule or outside of the molecule?
- 5. Two intermolecular forces exist in Figure 1. Where are they positioned relative to the molecules – within the molecules or between the molecules?



State the difference between intermolecular and intramolecular forces in terms of where they occur on the molecular level.

Teacher Key To Pogil Intermolecular Forces

Geoffrey C. Maitland

Teacher Key To Pogil Intermolecular Forces:

Chemists' Guide to Effective Teaching Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry Forces Pierre L. Huyskens, Werner A.P. Luck, Therese Zeegers-Huyskens, 2012-12-06 The study of intermolecular forces began over one hundred years ago in 1873 with the famous thesis of van der Waals In recent decades knowledge of this field has expanded due to intensive research into both its theoretical and the experimental aspects This is particularly true for the type of very strong cohesive force stressed in 1920 by Latimer and Rodebush the hydrogen bond a phenomenon already outlined in 1912 by Moore and Winemill Hydrogen bonds exert a profound influence on most of the physical and chemical properties of the materials in which they are formed Not only do they govern viscosity and electrical conductivity they also intervene in the chemical reaction path which determines the kinetics of chemical processes. The properties of chemical substances depend to a large extent on intermolecular forces. In spite of this fundamental fact too little attention is given to these properties both in research and in university teaching For instance in the field of pharmaceutical research about 13000 compounds need to be studied in order to find a single new product that can be successfully marketed. The recognition of the need to optimize industrial research efficiency has led to a growing interest in promoting the study of inter molecular forces Rising salary costs in industry have encouraged an interest in theoretical ideas which will lead to tailor made materials

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Geoffrey C. Maitland, 1981 Intermolecular Forces, Theory of Intermolecular Forces Henry Margenau, N. R. Kestner, 1969 Intermolecular Forces A. Pullman, 2014-01-15 Intermolecular Forces Taro Kihara, 1976

The book delves into Teacher Key To Pogil Intermolecular Forces. Teacher Key To Pogil Intermolecular Forces is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Teacher Key To Pogil Intermolecular Forces, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Teacher Key To Pogil Intermolecular Forces
 - Chapter 2: Essential Elements of Teacher Key To Pogil Intermolecular Forces
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 - Chapter 4: Teacher Key To Pogil Intermolecular Forces in Specific Contexts
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- 2. In chapter 1, the author will provide an overview of Teacher Key To Pogil Intermolecular Forces. The first chapter will explore what Teacher Key To Pogil Intermolecular Forces is, why Teacher Key To Pogil Intermolecular Forces is vital, and how to effectively learn about Teacher Key To Pogil Intermolecular Forces.
- 3. In chapter 2, this book will delve into the foundational concepts of Teacher Key To Pogil Intermolecular Forces. The second chapter will elucidate the essential principles that need to be understood to grasp Teacher Key To Pogil Intermolecular Forces in its entirety.
- 4. In chapter 3, this book will examine the practical applications of Teacher Key To Pogil Intermolecular Forces in daily life.

 This chapter will showcase real-world examples of how Teacher Key To Pogil Intermolecular Forces can be effectively utilized in everyday scenarios.
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- 6. In chapter 5, this book will draw a conclusion about Teacher Key To Pogil Intermolecular Forces. This chapter will summarize the key points that have been discussed throughout the book.

 The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Teacher Key To Pogil Intermolecular Forces.

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