

Assignment #2 – Compound Names and Formulas

Multi-valent ions only

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg₂O
3. gold (III) chloride AuCl₃
4. thallium (III) bromide TlBr₃
5. bismuth (V) oxide Bi₂O₅
6. terbium (III) oxide Tb₂O₃
7. uranium (VI) oxide UO₃
8. protactinium (IV) bromide PaBr₄
9. cerium (III) oxide Ce₂O₃
10. arsenic (V) sulphide As₂S₅

11. manganese (III) oxide Mn₂O₃
12. vanadium (II) bromide VBr₂
13. niobium(V) oxide Nb₂O₅
14. titanium (III) oxide Ti₂O₃
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co₃P₂
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr₂
20. copper (II) bromide CuBr₂

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl₄ Tin (IV) Chloride
2. BiBr₅ Bismuth (V) Bromide
3. PoO₂ Polonium (IV) Oxide
4. PbI₂ Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au₂O₃ Gold (III) Oxide
8. FeCl₂ Iron (II) Chloride

9. PdF₄ Palladium (IV) Fluoride
10. Os₂O₃ Osmium (III) Oxide
11. MoBr₂ Molybdenum (II) Bromide
12. VCl₅ Vanadium (V) Chloride
13. Mn₂O₃ Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np₂O₃ Neptunium (III) Oxide
16. V₂O₅ Vanadium (V) Oxide

Teacher Answer Key Chapter 6 Names And Formulas

George E. Pawlas, Peter F. Oliva



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Table of Contents Teacher Answer Key Chapter 6 Names And Formulas

1. Understanding the eBook Teacher Answer Key Chapter 6 Names And Formulas
 - The Rise of Digital Reading Teacher Answer Key Chapter 6 Names And Formulas
 - Advantages of eBooks Over Traditional Books
2. Identifying Teacher Answer Key Chapter 6 Names And Formulas
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Teacher Answer Key Chapter 6 Names And Formulas
 - User-Friendly Interface
4. Exploring eBook Recommendations from Teacher Answer Key Chapter 6 Names And Formulas
 - Personalized Recommendations
 - Teacher Answer Key Chapter 6 Names And Formulas User Reviews and Ratings
 - Teacher Answer Key Chapter 6 Names And Formulas and Bestseller Lists

5. Accessing Teacher Answer Key Chapter 6 Names And Formulas Free and Paid eBooks
 - Teacher Answer Key Chapter 6 Names And Formulas Public Domain eBooks
 - Teacher Answer Key Chapter 6 Names And Formulas eBook Subscription Services
 - Teacher Answer Key Chapter 6 Names And Formulas Budget-Friendly Options
6. Navigating Teacher Answer Key Chapter 6 Names And Formulas eBook Formats
 - ePub, PDF, MOBI, and More
 - Teacher Answer Key Chapter 6 Names And Formulas Compatibility with Devices
 - Teacher Answer Key Chapter 6 Names And Formulas Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Teacher Answer Key Chapter 6 Names And Formulas
 - Highlighting and Note-Taking Teacher Answer Key Chapter 6 Names And Formulas
 - Interactive Elements Teacher Answer Key Chapter 6 Names And Formulas
8. Staying Engaged with Teacher Answer Key Chapter 6 Names And Formulas
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Teacher Answer Key Chapter 6 Names And Formulas
9. Balancing eBooks and Physical Books Teacher Answer Key Chapter 6 Names And Formulas
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Teacher Answer Key Chapter 6 Names And Formulas
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Teacher Answer Key Chapter 6 Names And Formulas
 - Setting Reading Goals Teacher Answer Key Chapter 6 Names And Formulas
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Teacher Answer Key Chapter 6 Names And Formulas
 - Fact-Checking eBook Content of Teacher Answer Key Chapter 6 Names And Formulas
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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