

Reviewing Content

42. The solvent is the substance in which the solute is dissolved.
43. Random collisions of the solvent molecules with the solute particles provide enough force to overcome gravity.
44. **solubility**: the amount of a substance that dissolves in a given quantity of solvent at specified conditions of temperature and pressure to produce a saturated solution.
saturated solution: a solution containing the maximum amount of solute for a given amount of solvent at a constant temperature and pressure. **unsaturated solution**: a solution that contains less solute than a saturated solution at a given temperature and pressure. **miscible**: describes liquids that dissolve in each other. **immiscible**: describes liquids that are insoluble in each other.
45. Particles of solute crystallize.
46. No; if there were undissolved solute, the excess solute would come out of a supersaturated solution.
47. 5.55×10^2 g AgNO_3
48. Solubility increases with pressure.
49. a. 1.6×10^{-2} g/L
b. 4.7×10^{-2} g/L
50. *Dilute* and *concentrated* are relative terms and are not quantitative. Molarity provides the exact number of moles of solute per liter of solution.
51. Molarity is the number of moles of solute dissolved in one liter of solution.
a. 1.3M KCl
b. 3.3×10^{-1} M MgCl_2
52. 2.00×10^3 mL
53. a. 5.0×10^{-3} mol NaCl, 29 g NaCl
b. 1.0 mol KNO_3 , 1.0×10^{-2} g KNO_3
c. 2.5×10^3 mol CaCl_2 , 2.8 g CaCl_2
54. a. 2.3×10^1 g NaCl
b. 2.0 g MgCl_2
55. a. 16% (v/v) ethanol
b. 63.6% (v/v) isopropyl alcohol
56. Colligative properties are properties of a solution that depend only on the number of solute particles; boiling-point elevation, freezing-point depression, and vapor-pressure lowering. Boiling points are elevated because shells of solvent form around solute particles, reducing the amount of solvent molecules that have sufficient energy to escape the solution; relative to the pure solvent, the amount of energy required to cause vaporization or boiling increases. Solutes disrupt the ordering of the solvent structure, so more kinetic energy must be withdrawn from a solution for it to solidify. This lowers the freezing point of the solution.
57. a. sea water
b. 1.5M KNO_3
c. 0.100M MgCl_2
58. The effective molality of the $\text{Ca}(\text{NO}_3)_2$ solution is 3m. The effective molality of the NaNO_3 solution is 2m.
59. When vapor pressure is lowered relative to pure solvent, more energy must be supplied to reach the boiling point; thus the boiling point is increased relative to pure solvent.
60. The salt lowers the freezing point of the ice-water cooling mixture.
61. 1M solution: 1 mol of solute in 1 L of solution; 1m solution: 1 mol of solute in 1000 g of solvent
62. Add 27.0 g H_2O to 32.0 g CH_3OH .
63. a. 100.26°C
b. 101.54°C
64. a. -4.46°C
b. -2.2°C
65. a. -1.1°C
b. -0.74°C
c. -1.5°C

Understanding Concepts

66. a. The freezing-point depression is twice as great for solute B; solute B must provide twice as many particles in solution.

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JL Elias

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Can Potatoes Feed the World? John E. Bradshaw,2025-07-01 By 2050 the United Nations UN predicts a world population of 9.7 billion compared with 8 billion in 2022 Increases in food and energy production and the supply of fresh water will be needed to sustain this population whilst reducing greenhouse gas emissions to limit global warming The average global temperature is likely to be 2 to 3 degrees Celsius above that in pre industrial times unless there is a greater sense of urgency following the UN Climate Change Conference COP26 in Glasgow in 2021 which still wanted to limit the rise to 1.5 degrees There is also increasing concern about the loss of biodiversity on Earth from human activity including farming as seen in the outcomes of the UN Biodiversity Conference COP15 in Montreal in 2022 More encouragingly on 1 December 2023 134 countries at COP28 in Dubai endorsed the landmark sustainable agriculture resilient food systems and climate action declaration which put food systems transformation on the global climate agenda While food systems are vital for meeting societal needs and enabling adaptation to climate impacts they are also responsible for as much as a third of global greenhouse gas emissions A warmer climate and loss of biodiversity will make life more difficult for humankind but can potatoes at least help with food security It is a scientific and technological question set in a political economic and societal context It has arisen because potatoes have made the journey from wild species to global food crop The contribution of the potato to the United Nations Millennium Development Goals of providing food security and eradicating poverty was recognized when an International Year of the Potato 2008 IYP 2008 was officially launched at the United Nations UN headquarters in New York on 18 October 2007 by the Director General of FAO Food and Agriculture Organization of the

United Nations Today the importance of potatoes can be seen in the context of the United Nations 2030 Agenda for Sustainable Development which was adopted in 2015 and started on 1 January 2016 The agenda has 17 goals the second of which SDG2 is to end hunger achieve food security and improved nutrition and promote sustainable agriculture By 2030 the aim of the agenda is to ensure access by all people in particular the poor and people in vulnerable situations including infants to safe nutritious and sufficient food all year round However the projection in the 2023 Sustainable Development Goals Report was not zero hunger but 600 million people still facing hunger Hence a greater sense of urgency is required and also a need to look beyond 2030 to 2050 This book explores how potatoes can contribute to SDG2 by increasing potato production and improving the nutritional value of potatoes in particular to alleviate micronutrient deficiencies hidden hunger having first explained how potatoes became a major food crop and the lessons to be learnt from a major crop failure and resulting famine

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